



Contents lists available at ScienceDirect

Journal of the European Ceramic Society

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## Original Article

Structure, microwave dielectric properties, and infrared reflectivity spectrum of olivine type  $\text{Ca}_2\text{GeO}_4$  ceramicYing Tang<sup>a,b</sup>, Minyu Xu<sup>b</sup>, Lian Duan<sup>b</sup>, Junqi Chen<sup>b</sup>, Chunhui Li<sup>c</sup>, Huaicheng Xiang<sup>b</sup>,  
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## ARTICLE INFO

## Keywords:

Olivine structure

 $\text{Ca}_2\text{GeO}_4$  ceramicLow- $\epsilon_r$ High  $Q \times f$ 

Infrared reflectivity spectrum

## ABSTRACT

$\text{Ca}_2\text{GeO}_4$  dielectric ceramic was prepared using the conventional solid-state reaction method. Sintering behavior, crystal structure, microstructure, and microwave dielectric properties were analyzed by XRD, SEM, Raman, and Infrared reflectivity spectrum.  $\text{Ca}_2\text{GeO}_4$  was found to crystallize in the olivine structure with a space group of  $Pnma$ . A dense and high-performance microwave dielectric property with permittivity  $\sim 6.76 \pm 0.02$ ,  $Q \times f$  value  $\sim 82,400 \pm 1800$  GHz, and temperature coefficient  $\sim -67 \pm 3.4$  ppm/ $^\circ\text{C}$  were obtained in the sample sintered at  $1420^\circ\text{C}$ . Infrared spectral analysis supported that the dielectric contribution for  $\text{Ca}_2\text{GeO}_4$  at microwave region is dominated by absorption of phonons and there is no contribution from dipolar or other polarization mechanisms. The large negative  $\tau_f$  values could be compensated by forming composite ceramics with  $\text{CaTiO}_3$ . A low- $\epsilon_r$  of  $9.02 \pm 0.03$ , a high  $Q \times f$  of  $49,880 \pm 1400$  GHz, and a near-zero  $\tau_f$  value of  $+4 \pm 0.6$  ppm/ $^\circ\text{C}$  were obtained for  $0.92\text{Ca}_2\text{GeO}_4\text{-}0.08\text{CaTiO}_3$  ceramic at  $1420^\circ\text{C}$  for 4 h. This ceramic could be a good candidate for microwave substrate materials.

## 1. Introduction

Microwave dielectric ceramics as a kind of core materials for microwave devices (such as resonators, filters, substrates etc.) are widely used in intelligent transportation systems (ITS), radar, mobile phone and global positioning system (GPS) etc [1–3]. Commonly, the judgments of microwave properties have three prime requirements: the suitable dielectric constant  $\epsilon_r$ , the high quality factor  $Q$ , and the stable temperature coefficient of resonant frequency ( $\tau_f$ ) [4,5]. The era of miniaturization, high frequency and integration of microwave dielectrics is coming. It means more powerful microwave substrates should be exploited, which could carry more components and exhibit better dielectric and thermal properties. Since the wavelength inside the devices is inversely proportional to the square root of its dielectric constant, the use of high  $\epsilon_r$  materials can be beneficial to miniaturization, but high dielectric constant materials commonly increase the leakage current of integrated circuits, the capacitance effect between wires and the heating of integrated circuits. Nevertheless, the time of signal delay is proportional to the square root of dielectric constants, and the low- $\epsilon_r$  materials could enhance the transition speed, and decrease the heating and system loss during high working frequency

transmission [6–10]. Thus, the low- $\epsilon_r$ , high- $Q$  and thermal stability materials could be considered to be the most promising candidates for the next-generation substrates in high frequency electronic devices.

Up to now, the typical low- $\epsilon_r$  and high- $Q$  microwave dielectric materials including Al-, Si-based materials, such as spinel structure  $\text{MAl}_2\text{O}_4$  ( $\text{M} = \text{Zn, Mg, Ni, Co}$ ) [11–13], olivine structure  $\text{A}_2\text{SiO}_4$  ( $\text{A} = \text{Ca, Mg}$ ) [14,15]. But, the sintering temperatures of Al-containing compounds are suitable to co-fired with the refractory metal electrode, such as W, Mo, which should be sintered with ceramics in the range of  $1200\text{--}1600^\circ\text{C}$  with the reducing atmosphere, resulting in a more expensive cost and more difficult process. During these materials, some Si-containing olivine structure dielectric materials have attracted considerable attention, such as  $\text{Mg}_2\text{SiO}_4$  ( $\epsilon_r = 7.5$ ,  $Q \times f = 114,730$  GHz,  $\tau_f = -57$  ppm/ $^\circ\text{C}$ ) [15]. As well-known, germanium and silicon are in the same group in the periodic table, and Ge atom has the same electrovalence and similar electronic shell structure to Si atom. Based on this, Chen et al. [16] first reported the  $\text{Mg}_2\text{GeO}_4$  dielectric material with low- $\epsilon_r$  and high- $Q$ . Furthermore, Eulenberger et al. reported the synthesis of olivine structure  $\text{Ca}_2\text{GeO}_4$  in their optical studies and used it in laser material [17–19]. To date, the microwave dielectric properties of  $\text{Ca}_2\text{GeO}_4$  have not been studied yet. Considering the excellent

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Received 19 December 2018; Received in revised form 16 February 2019; Accepted 19 February 2019

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dielectric performances of  $\text{Mg}_2\text{GeO}_4$ , thus, it is reasonable to predict that  $\text{Ca}_2\text{GeO}_4$  might possess promising microwave dielectric properties, and it is also worth to investigate the relationship between structure and microwave properties.

In this work,  $\text{Ca}_2\text{GeO}_4$  ceramics were synthesized by the traditional solid-state reaction method. The relationships between sintering behavior, crystal structure, microstructure, and microwave dielectric properties were analysis by XRD, SEM, Raman, and infrared reflectivity spectrum in detail.

## 2. Experimental section

$\text{Ca}_2\text{GeO}_4$  ceramics were prepared by a conventional solid-state reaction method using high-purity powders of  $\text{CaCO}_3$  (99.99%, Guo-Yao Co. Ltd., China) and  $\text{GeO}_2$  (99.99%, Guo-Yao Co. Ltd., China) as the raw materials. The weighted powers were ball-milled for 4 h, dried, and calcined at 1250 °C for 4 h. The calcined powders were re-milled for 4 h with PVA as binders. The granulated powders were then pressed into cylinders (10 mm in diameter and 6 mm in thickness) in a steel die under uniaxial pressure of 200 MPa. The samples were heated at 550 °C for 4 h to remove the organic binders and sintered at 1350–1440 °C for 4 h at a rate of 5 °C/min and naturally cooled to the room temperature. The thin pellet was prepared for room temperature infrared reflectivity spectra measurement.

The relative densities of the sintered samples were measured with the Archimede's method. The phase purity of the fired ceramics was investigated using an X-ray diffractometer (XRD; Model X'Pert PRO, PANalytical, Almelo, the Netherlands). The microstructures of the samples were examined by field-emission scanning electron microscopy (FESEM; S4800, Hitachi, Tokyo, Japan). The Raman spectra at room temperature were obtained with a Raman spectrometer (DXR; Thermo Fisher Scientific, American). Microwave dielectric properties of the sintered ceramics were measured using a network analyzer (N5230 A, Agilent Co., Palo Alto, California) and a temperature chamber (Delta 9039, Delta Design, San Diego, California). The  $\tau_f$  values were obtained in the temperature range of 25 °C–85 °C:

$$\tau_f = \frac{f_2 - f_1}{f_1(T_2 - T_1)} \quad (1)$$

where,  $f_1$  and  $f_2$  represent resonant frequencies at temperatures  $T_1$  and  $T_2$ , respectively.

The room-temperature infrared reflectivity spectra of the pellet with the surface carefully polished were measured using a Bruker IFS 66v FTIR spectrometer on Infrared beamline station at National Synchrotron Radiation Lab. (NSRL), China. The resolution of the spectrum was selected as 2  $\text{cm}^{-1}$  and the incident angle of radiation was fixed at 12°. The spectrum in the frequency range of 50–4000  $\text{cm}^{-1}$  was obtained by the combination of the two spectra in the frequency ranges of 50–600 and 400–4000  $\text{cm}^{-1}$ . In the former range, an FIR-DTGS far-infrared-deuterated triglycine sulfate detector and Mylar multilayer film beam splitter (total thickness of 6  $\mu\text{m}$ ) were employed, while in the latter range, the FIR-DTGS detector and Ge/KBr beam splitter were employed. The far-infrared reflection spectra of  $\text{Ca}_2\text{GeO}_4$  ceramics were collected in the wavenumber range from 50  $\text{cm}^{-1}$  to 1000  $\text{cm}^{-1}$ , and were analyzed using a classical harmonic oscillator model (the Lorentz three parameter classical model) [20]:

$$\varepsilon^*(\omega) = \varepsilon_\infty + \sum_{j=1}^n \left( \frac{\omega_{pj}^2}{\omega_{oj}^2 - \omega^2 - j\gamma_j\omega} \right) \quad (2)$$

where  $\varepsilon_\infty$  is the dielectric constant caused by the electronic dispersive polarization at optical frequencies;  $\gamma_j$ ,  $\omega_{oj}$ , and  $\omega_{pj}$  are the damping factor, the transverse frequency, and plasma frequency of the  $j$ th Lorentz oscillator, respectively, and  $n$  is the number of transverse phonon modes. The contribution to dielectric permittivity of each oscillator  $\Delta\varepsilon_j$  i.e. strength is calculated as the ratio of  $(\omega_{pj})^2$  to  $(\omega_{oj})^2$ . The reflectivity

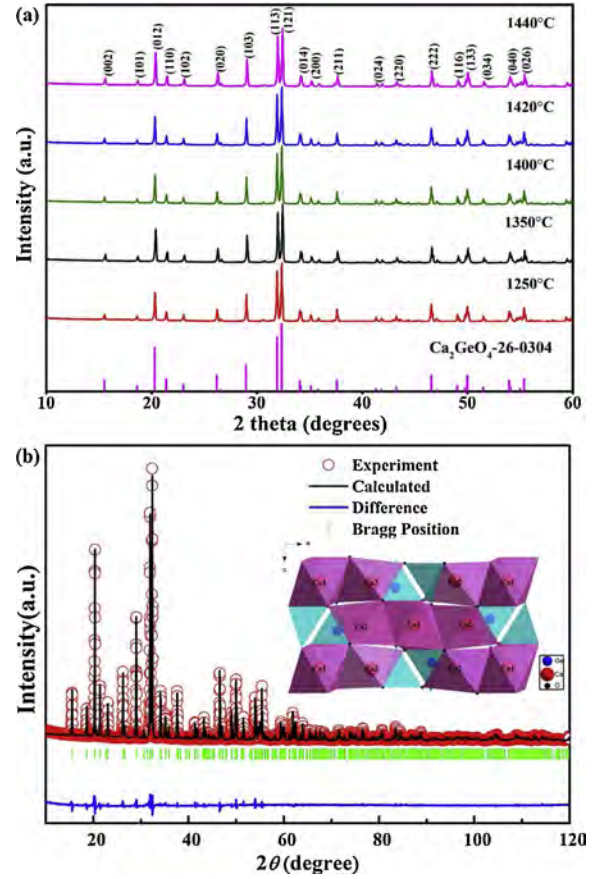


Fig. 1. (a) The XRD patterns of  $\text{Ca}_2\text{GeO}_4$  powder prepared at 1250 °C and sintered at 1350–1440 °C; (b) the refinement of the  $\text{Ca}_2\text{GeO}_4$  ceramics sintering at 1420 °C. The inset is the crystal structure of  $\text{Ca}_2\text{GeO}_4$ .

$R(\omega)$  can be calculated from the complex dielectric function as according to [21]:

$$R(\omega) = \left| \frac{1 - \sqrt{\varepsilon^*(\omega)}}{1 + \sqrt{\varepsilon^*(\omega)}} \right|^2 \quad (3)$$

## 3. Results and discussion

Fig. 1 shows the XRD patterns of  $\text{Ca}_2\text{GeO}_4$  prepared and sintered at different sintering temperatures (1250–1440 °C). By indexing with the standard PDF card (#26-0304), no additional peaks were detected at 1250 °C. All existing peaks in samples sintered at 1350, 1400, 1420 and 1440 °C are related to  $\text{Ca}_2\text{GeO}_4$  (PDF card #26-0304) without any additional phases. Based on the results of phase analysis, Rietveld refinement was performed on the sintered sample at 1420 °C using the Fullprof software with olivine-type  $\text{Ca}_2\text{GeO}_4$  as the structural model. And  $\text{Ca}_2\text{GeO}_4$  (ICSD #26-0304) as a starting model for the Rietveld refinement, the symbols represent the experimental data and the solid line could be satisfactorily fitting to the diffraction data which suggested the  $\text{Ca}_2\text{GeO}_4$  powders (sintering at 1420 °C) crystallized in an orthorhombic olivine structure with a space group of  $Pnma$ ,  $a = 11.3997 \text{ \AA}$ ,  $b = 6.7919 \text{ \AA}$ ,  $c = 5.2400 \text{ \AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ , and acceptable  $R_p = 4.11\%$ ,  $R_{wp} = 5.82\%$ , and  $R_{exp} = 3.16\%$ ,  $\chi^2 = 3.39$ . The inset of Fig. 1(b) was the olivine crystal structure for  $\text{Ca}_2\text{GeO}_4$  ceramic. All the ions adopt the olivine structure and the unit cell of  $\text{Ca}_2\text{GeO}_4$  consists of nine  $\text{CaO}_6$  octahedral and six  $\text{GeO}_4$  tetrahedral [22].

Fig. 2 shows the SEM images of the  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at various temperatures. At low sintering temperature (1350 °C) the

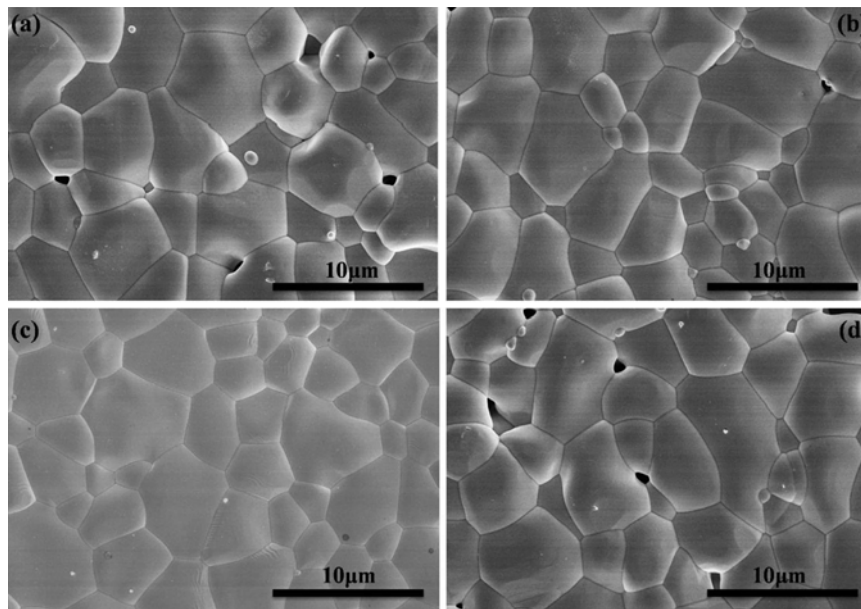


Fig. 2. The SEM images of the  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at different temperatures: (a) 1350 °C; (b) 1400 °C; (c) 1420 °C; (d) 1440 °C.

**Table 1**

The microwave dielectric properties ( $\epsilon_r$ ,  $Q \times f$ , and  $\tau_f$ ), bulk density and FWHM of the  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at 1350–1440 °C.

S.T. (°C)	$\epsilon_r$	$Q \times f$ (GHz)	$\tau_f$ (ppm/°C)	Density (g/cm <sup>3</sup> )	Relative density (%)	FWHM
1350	$6.60 \pm 0.05$	$58,100 \pm 2,900$	$-66 \pm 2.7$	$3.15 \pm 0.03$	89.2	6.03
1400	$6.67 \pm 0.02$	$74,630 \pm 2,700$	$-65 \pm 3.2$	$3.30 \pm 0.02$	93.4	5.99
1420	$6.76 \pm 0.02$	$82,400 \pm 1800$	$-67 \pm 3.4$	$3.35 \pm 0.02$	94.9	5.68
1440	$6.59 \pm 0.03$	$67,900 \pm 2,000$	$-66 \pm 2.5$	$3.28 \pm 0.02$	92.9	6.19

$\text{Ca}_2\text{GeO}_4$  ceramic exhibited a small amount of pores. However, when the sintering temperature increased to 1420 °C, the  $\text{Ca}_2\text{GeO}_4$  ceramic exhibited a relatively dense microstructure and few visible pores (Fig. 2(c)). As the sintering temperature increased to 1440 °C, abnormal large grains (~11 μm) grains were overgrown and some pores re-occurred, which due to the excessive sintering temperature. The bulk densities of the sintered samples are listed in Table 1. As the sintering temperature increased, the bulk density just slightly increased from 3.15 g/cm<sup>3</sup> at 1350 °C, to a maximum value of 3.35 g/cm<sup>3</sup> at 1420 °C (~94.9% of the theoretical density), and then decreased to 3.28 g/cm<sup>3</sup> at 1440 °C. This decline in density might be ascribed to the formation of pores associated with several abnormal grain growths and a small amount of porosity at 1440 °C. Table 1 also lists the microwave dielectric properties ( $\epsilon_r$ ,  $Q \times f$ , and  $\tau_f$ ) of the  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at 1350–1440 °C. The optimum microwave dielectric properties with  $\epsilon_r = 6.76 \pm 0.02$ ,  $Q \times f = 82,400 \pm 1800$ , and  $\tau_f = -67 \pm 3.4$  ppm/°C were obtained when sintered at 1420 °C. With the variation of sintering temperatures, microwave dielectric properties and densities have a similar trend, where the maximum values were obtained correspondingly with the maximal densities. It is well known that the  $\epsilon_r$  was depended on density, ionic polarizability, and secondary phase. Considering the effect of density, the porosity-corrected permittivity ( $\epsilon_{\text{corrected}}$ ) was calculated by the Bosman-Having's equation [23].

$$\epsilon_{\text{corrected}} = \epsilon_r (1 + 1.5p) \quad (4)$$

where,  $p$  is the fractional porosity. The  $\epsilon_{\text{corrected}}$  value for  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at 1420 °C is 7.34, which is slightly higher than the calculated theoretical permittivity  $\epsilon_{\text{theo}}$  (6.83) of  $\text{Ca}_2\text{GeO}_4$  calculated by the Clausius-Mosotti equation [24].

Usually,  $Q \times f$  values would be influenced by the extrinsic and intrinsic factors in microwave frequency range. The extrinsic losses are

mainly attributed to the lattice defects, including second phases, oxygen vacancies, grain boundaries, and densification or porosities, which could be effectively reduced by optimizing the preparation parameters. According to the XRD results, the effects of the secondary phase could be neglected due to the absence of detection of secondary phases, and the variation of densification of  $\text{Ca}_2\text{GeO}_4$  with sintering temperature has the similar change tendency as  $Q \times f$  values, thus, sintering temperature and densification might be the main extrinsic reason for the change of  $Q \times f$  value in  $\text{Ca}_2\text{GeO}_4$  ceramics.

Some  $\text{A}_2\text{BO}_4$ -type olivine ceramics and their microwave dielectric properties are given in Table 2. As can be seen, these  $\text{A}_2\text{BO}_4$ -type dielectric ceramics have some similar characteristics, such as high sintering temperature, high  $Q \times f$ , low  $\epsilon_r$ , and negative  $\tau_f$ . Comparing to other olivine structure materials,  $\text{Ca}_2\text{GeO}_4$  has a much higher  $Q \times f$  value than that of  $\text{Li}_2\text{A}'\text{BO}_4$  ( $\text{A}' = \text{Zn, Mg; B} = \text{Ge, Si}$ ), but slightly lower than  $\text{Mg}_2\text{SiO}_4$  and  $\text{Mg}_2\text{GeO}_4$  [10,14–16,25–27]. The property difference might be due to the local structure variance induced by different atomic

**Table 2**

Microwave dielectric properties of some  $\text{A}_2\text{BO}_4$ -type ceramics with olivine structure.

Ceramics	S.T. (°C)	$\epsilon_r$	$Q \times f$ (GHz)	$\tau_f$ (ppm/°C)	Reference
$\text{Li}_2\text{MgSiO}_4$	1250	5.1	15,380	/ <sup>a</sup>	[25]
$\text{Li}_2\text{MgGeO}_4$	1220	6.1	28,500	-74.7	[10]
$\text{Ca}_2\text{GeO}_4$	1420	6.76	82,400	-67	This work
$\text{Li}_2\text{ZnGeO}_4$	1200	6.5	35,400	-60.6	[10]
$(\text{Zn}_{0.95}\text{Co}_{0.05})_2\text{SiO}_4$	1200	6.5	57,000	-55.0	[26]
$\text{Mg}_2\text{GeO}_4$	1250	6.76	95,000	-28.7	[16]
$\text{Mg}_2\text{SiO}_4$	1550	7.5	114,730	-59	[15]
$\text{Mg}_2\text{SnO}_4$	1600	8.41	55,100	-62	[27]
$\text{Ca}_2\text{SiO}_4$	1450	10.1	26,000	-89	[14]

<sup>a</sup> Not studied.



occupation [28,29]. Redhammer et al. [22]. have reported that the different size of the M1 cation (this position is equal to Ca1 in  $\text{Ca}_2\text{GeO}_4$ ) might change the polyhedral distortion, expressed by the parameters bond-length distortion, octahedral angle variance, and octahedral quadratic elongation, and the Ca germanate olivine compounds generally have more regular octahedra than the analogous silicates. Moreover, the intrinsic loss is sensitive to crystal structure, which is affected by the vibration mode of the lattice, packing fraction, magnetic loss, and structure tilting [30,31]. Therefore, the intrinsic dielectric loss of  $\text{Ca}_2\text{GeO}_4$  would be explored by the Raman spectrum and infrared reflectivity spectrum in the following part.

In general, the longer dimensional range associated with diffraction phenomena restricts the prediction of the variation in the local bonding characteristics through powder X-ray diffraction measurements. The Raman spectrum is very sensitive to the nature of bonding, and it is helpful to identify the characterizing of the local structure. According to the group theoretical method [32], there are 36 Raman active optical phonon modes, but the  $\text{Ca}_2\text{GeO}_4$  ceramics observed in the Raman spectra are only 16–22 peaks, which is less than the theoretically expected Raman active modes. This phenomenon might be arising from the mutual influence of the mixed or overlapped Raman active vibration modes and resolution of the measuring instrument as well as the partial symmetry of the sample being destroyed, and many modes cannot be separated from one another at the spectral resolution employed.

The room-temperature Raman spectra of  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at different temperatures (1350, 1400, 1420, and 1440 °C) are presented in Fig. 3. The Raman in the range of 600–1000  $\text{cm}^{-1}$  are assigned to stretching modes of the  $\text{GeO}_4$  tetrahedra in the structure, and the strongest peak was assigned to the stretching vibration of  $\text{GeO}_4$  tetrahedron [33], which was actually the symmetric Ge-O stretching mode ( $A_g$  symmetry) in  $\text{GeO}_4$  tetrahedron; the bands observed in the range of medium range were associated with antisymmetric bending vibrations of  $\text{GeO}_4$  tetrahedra. And the mode below 200  $\text{cm}^{-1}$  is associated with the translational lattice motions of the  $\text{Ca}^{2+}$  cations [34]. Bending vibrations would be expected at lower frequency, but separating these from vibrations involving cations or lattice modes is difficult because of the occurrence of modes coupling and modes.

For a Raman mode, the FWHM is inversely correlated to the lifetime of the phonon, and usually, the narrower peak corresponds to the longer lifetime of phonon and less interaction with phonons [35]. These interactions would consume more energy, thereby decreasing the intrinsic dielectric loss and leading the microwave ceramic shows a low dielectric loss. The FWHM of  $\text{Ca}_2\text{GeO}_4$  ceramic at the strongest Raman mode are listed in Table 1, the relative lower FWHM of Ge-O stretching mode with have a relative dielectric loss, which has a similar rule as the  $Q \times f$  value.

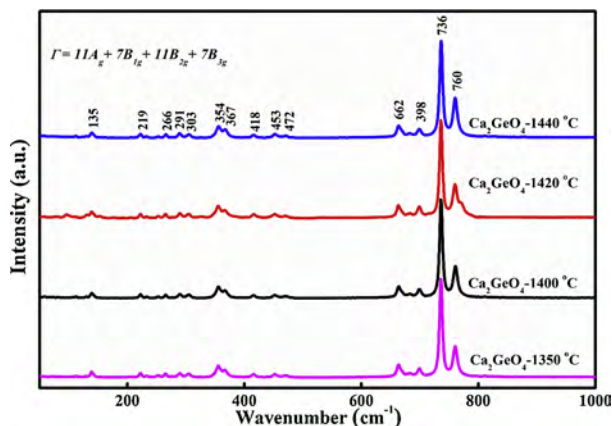


Fig. 3. The room-temperature Raman spectra of  $\text{Ca}_2\text{GeO}_4$  ceramics sintered at different temperatures (1350–1440 °C).

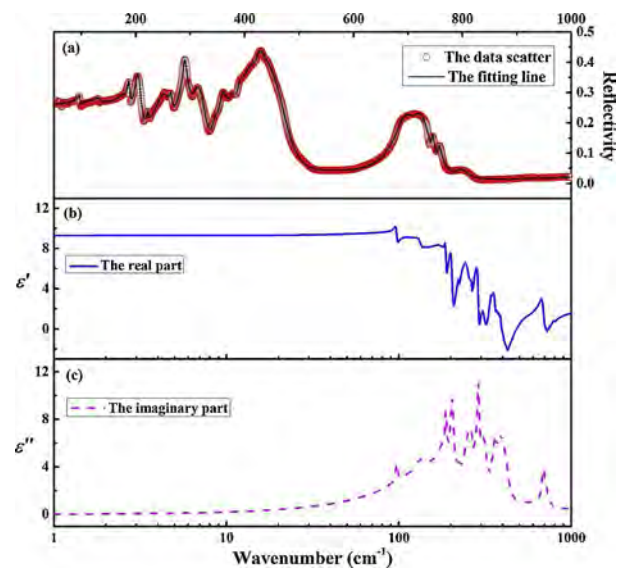


Fig. 4. The measured and calculated infrared reflectivity spectra of the  $\text{Ca}_2\text{GeO}_4$  ceramics, (a) black solid lines are fitting values and red circles for measured values; (b) the blue solid is the real part; the pink line is the imaginary part. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article).

To further study the intrinsic microwave dielectric properties, the analysis based on infrared reflectivity spectra of the  $\text{Ca}_2\text{GeO}_4$  were necessary as the infrared reflection spectra are mainly caused by the absorption of the polar lattice vibration. All calculated IR reflectivity spectrum of  $\text{Ca}_2\text{GeO}_4$  was fitted well by 24 resonant modes though Eqs. (2)–(3), it is difficult to obtain all the independent peaks due to the instrumental resolution and interactions among these modes. The fitted spectra are in good agreement with the experimental ones, all of these data could be observed in Fig. 4 and Table 3. The spectrum data in the ranges of 0–50  $\text{cm}^{-1}$  and 1000–5000  $\text{cm}^{-1}$  without IR peaks were not considered to avoid instability, which may be caused by multiple reflections inside the grains and other unexpected effects. In order to represent more details, only the region with the frequency below

Table 3  
The fitting parameters of vibration modes of  $\text{Ca}_2\text{Ge}_2\text{O}_4$ .

$\omega_{oj}(\text{cm}^{-1})$	$\omega_{oj}(\text{cm}^{-1})$	$\gamma_j$	$\Delta\epsilon_j$
97.089	21.907	3.2004	0.0509
160.81	290.73	152.47	3.27
181.04	110.67	30.992	0.374
187.38	49.857	4.1671	0.0708
203.43	114.07	10.473	0.314
222.34	28.875	3.6472	0.0169
253.25	197.95	33.58	0.611
264.17	42.804	4.9092	0.0263
286.87	168.91	20.791	0.347
288.78	78.698	6.3809	0.0743
312.05	188.1	29.338	0.363
345.86	65.116	11.133	0.0354
358.95	149.91	17.354	0.174
375.12	132.37	20.982	0.125
389.84	81.91	11.664	0.0441
403.31	288.97	46.059	0.513
422.51	66.272	15.001	0.0246
442.82	71.177	38.665	0.0258
576.57	254.13	213.82	0.194
688.59	293.5	40.664	0.182
715.29	140.52	33.251	0.0386
743.39	32.836	4.9484	0.00195
755.47	36.027	6.225	0.00227
796.17	73.868	24.626	0.00861
$\epsilon_\infty = 1.88$			

**Table 4**

Microwave dielectric properties of  $(1-x)\text{Ca}_2\text{GeO}_4\cdot x\text{CaTiO}_3$  composites sintered at different temperatures.

x values	S.T.(°C)	$\epsilon_r$	$Q \times f$ (GHz)	$\tau_f$ (ppm/°C)
0	1420	$6.76 \pm 0.02$	$82,400 \pm 1800$	$-67 \pm 3.4$
0.02	1390	$7.25 \pm 0.03$	$53,580 \pm 2,500$	$-45 \pm 2.5$
0.04	1400	$7.72 \pm 0.03$	$51,300 \pm 1,900$	$-23 \pm 1.5$
0.06	1410	$8.26 \pm 0.04$	$50,460 \pm 1,600$	$-10 \pm 1.3$
0.08	1420	$9.02 \pm 0.03$	$49,880 \pm 1400$	$+4 \pm 0.6$
0.10	1430	$9.43 \pm 0.05$	$48,040 \pm 1300$	$+25 \pm 1.5$

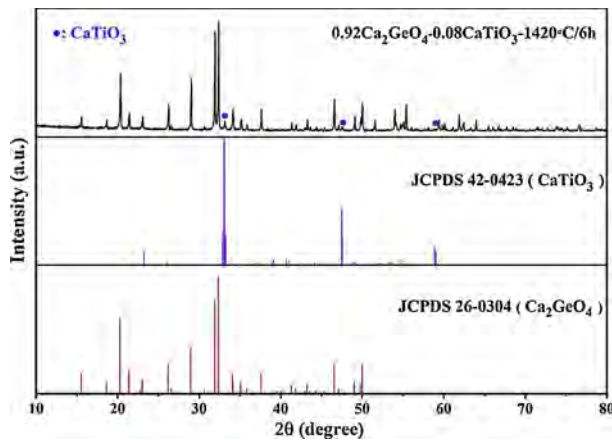


Fig. 5. The XRD patterns of  $0.92\text{Ca}_2\text{GeO}_4\cdot 0.08\text{CaTiO}_3$  ceramics.

$1000\text{ cm}^{-1}$  is displayed. The fitted parameters are listed in Table 3. The phonon parameters illustrate that the dielectric constant can reach 6.89 and contribute mainly (above 78.56%) to the dielectric polarization at microwave frequencies, which are below  $800\text{ cm}^{-1}$ , and the calculated  $Q \times f$  value was  $134,198\text{ GHz}$  using Eq. (4). But it is difficult to calculate every contribution to the polarization from each vibrational mode at microwave frequencies. It is seen that the calculated permittivity ( $\epsilon_r = 8.768$ ) are slightly larger than the experimental one ( $\epsilon_r = 6.76$ ) in the microwave region, it could be concluded that the polarization is dominated by absorption of phonons at the far infrared region and that there is no contribution from dipolar or other polarization mechanisms [36]. But the calculated  $Q \times f$  values is higher than the measured values, which means extrinsic loss and intrinsic loss might be influenced on  $Q \times f$  values in the microwave range, thus, it is possible to enhance the  $Q \times f$  values by optimizing the preparation processes.

In order to satisfy the requirements of electronic devices, microwave dielectric materials should have a near-zero  $\tau_f$  value. In general, two main methods are proposed to improve the  $\tau_f$  value. The first method is the formation of a composite between two compounds with opposite signs of  $\tau_f$  value, for example, the addition of  $\text{TiO}_2$  with  $\tau_f = +450\text{ ppm/}^\circ\text{C}$  can adjust the negative  $\tau_f$  such as in  $(1-x)\text{ZnZrNb}_2\text{O}_8\cdot x\text{TiO}_2$ ,  $(1-x)\text{ZnMoO}_4\cdot x\text{TiO}_2$  [37,38]. The second accepted method to obtain a near-zero  $\tau_f$  value is forming solid solution phases. This method is desirable owing to its ability to maintain the high  $Q \times f$  while successfully tuning the  $\tau_f$  value. More recently, the  $\text{CaTiO}_3$  ceramic with a large positive  $\tau_f \sim +800\text{ ppm/}^\circ\text{C}$  was prepared and characterized [39]. Therefore,  $\text{CaTiO}_3$  was used to improve the  $\tau_f$  values of  $\text{Ca}_2\text{GeO}_4$  ceramics. Table 4 shows the microwave dielectric properties of  $(1-x)\text{Ca}_2\text{GeO}_4\cdot x\text{CaTiO}_3$  ceramics sintered at different temperatures ( $1390\text{--}1430\text{ }^\circ\text{C}$ ). Temperature stable ceramics with a near-zero  $\tau_f$  value ( $+4 \pm 0.6\text{ ppm/}^\circ\text{C}$ ) for  $0.92\text{Ca}_2\text{GeO}_4\cdot 0.08\text{CaTiO}_3$  were obtained.

XRD pattern of  $0.92\text{Ca}_2\text{GeO}_4\cdot 0.08\text{CaTiO}_3$  ceramic is shown in Fig. 5. From the XRD patterns, only the peaks of  $\text{Ca}_2\text{GeO}_4$  and  $\text{CaTiO}_3$  (JCPDS No. 042-0423) could be detected without any other secondary phase, indicating that  $\text{Ca}_2\text{GeO}_4$  and  $\text{CaTiO}_3$  ceramics did not react at  $1420\text{ }^\circ\text{C}$ .

#### 4. Conclusion

In this work,  $\text{Ca}_2\text{GeO}_4$  ceramics with orthorhombic olivine structure were synthesized by the conventional solid-state reaction route. The optimum microwave dielectric properties of  $\text{Ca}_2\text{GeO}_4$  ceramics with  $\epsilon_r = 6.76 \pm 0.02$ ,  $Q \times f = 82,400 \pm 1800\text{ GHz}$ , and  $\tau_f = -67 \pm 3.4\text{ ppm/}^\circ\text{C}$  were obtained when sintered at  $1420\text{ }^\circ\text{C}$ . According to the result of infrared reflectivity, the polarization is dominated by absorption of phonons at the far infrared region, which is not affected by dipolar or other polarization mechanisms. The large negative  $\tau_f$  values could be compensated by forming composite ceramics, a near zero  $\tau_f$  values of  $+4 \pm 0.6\text{ ppm/}^\circ\text{C}$  were obtained for  $0.92\text{Ca}_2\text{GeO}_4\cdot 0.08\text{CaTiO}_3$  ceramic sintered at  $1420\text{ }^\circ\text{C}$  for 4 h. The olivine structure  $\text{Ca}_2\text{GeO}_4$  could be a promising candidate for microwave substrate materials.

#### Acknowledgments

This work was supported by Natural Science Foundation of China (Nos. 21761008 and 21561008), the Natural Science Foundation of Guangxi Zhuang Autonomous Region (Nos. 2016GXNSFBA380134, and 2015GXNSFFA139003, 2018GXNSFAA138175, 2018GXNSFAA281093), and Projects of Education Department of Guangxi Zhuang Autonomous Region (No. 2018KY0255). The authors would also like to thank the administrators in the IR beamline workstation of National Synchrotron Radiation Laboratory (NSRL) for their help in the IR measurements.

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